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Pullulan: Production and usage in food industry

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Pullulan is one of such polymers synthesized by the yeast-like fungus *Aureobasidium pullulans*. Pullulan is a linear α -D-glucan built of maltotriose subunits, connected by (1-6)- α -D-glycosidic linkages. Pullulan is an important exopolysaccharide having applications in several industrial sectors like pharmaceutical, food and cosmetic industries. In addition to recently pullulan is also being investigated for its biomedical applications in various aspects like targeted drug and gene delivery, tissue engineering, wound healing and in diagnostic imaging using quantum dots. Pullulan is being used extensively in the food industry as a food ingredient for over 20 years in Japan, and has Generally Regarded As Safe (GRAS) status in the USA. Pullulan, which is generally materialized with microbial origin in food the production has widely usage as a coating agent in food formulations and packaging industry owing to its numerous properties. Despite the large number of uses, some of the problems associated with fermentative production of pullulan are (i) the formation of a melanin pigment; (ii) the inhibitory effects caused by high sugar concentrations in the medium; and (iii) the high cost associated with pullulan precipitation and recovery. In this review, general properties of pullulan, pullulan production, usage of pullulan in food industry were examined.

Keywords: Pullulan, Food industry, *Aureobasidium pullulan*.

INTRODUCTION

Microbially produced polysaccharides have properties that are very useful in various industrial applications (Choudhury et al. 2011). A new fungal exopolysaccharides (EPSs) with interesting industrial properties are well known (Singh et al. 2008). Pullulan is an extracellular and neutral microbial polysaccharide produced by *Aureobasidium pullulans* in starch and sugar cultures (Cheng et al., 2010; Karim et al., 2009; Wu et al., 2012; Xiao et al., 2012 a,b). *A. pullulans* have five different cell morphologies like swollen blastospores, yeast-like cells, mycelia, chlamyospores and young blastospores (Ronen et al., 2002; Sugumaran et al., 2013). *A. pullulans* is a black yeast or yeast-like fungus widely spread in all ecological niches, e.g., forest soils, fresh and sea water, plant, and animal tissues, etc. (Leathers, 2003; Wu et al., 2012). It is a linear mixed linkage α -D-glucan consisting mainly of maltotriose repeating units interconnected by α -(1 \rightarrow 6) linkages. The regular alternation of α -(1 \rightarrow 4) and α -(1 \rightarrow 6) bonds results in structural flexibility and enhanced solubility (Bouvang

et al., 1963; Catley, 1970; Chi and Zao, 2003; Gniewosz and Duszkiwicz-Reinhard, 2008; Goksungur et al., 2011; Jakovljevic et al., 2001; Jiang et al., 2010, 2011; Leathers, 1993; McIntyre and Vogel, 1993; Roukas, 1998; Saha and Zeikus, 1989; Taguchi et al., 1973; Wallenfels et al., 1961; Wu et al., 2009a). Pullulan and its derivatives have numerous potential for food, pharmaceutical and other industrial applications (Chi and Zao, 2003; Deshpande et al., 1992; Gaur et al., 2010; Hannigan, 1984; Leathers 2003; Seviour et al., 1992; Shingel, 2004; Singh et al., 2008; Singh et al., 2010; Sutherland, 1998; Yuen, 1974). Pullulan can be used as coating and packaging material, sizing agent for paper, starch replacer in low-calorie food formulations, in cosmetics, lotions, shampoos and industrial applications (Deshpande et al., 1992; Leathers, 2003; Wu et al. 2009a). At present, the pullulan preparation process differs according to the intended application. For use as industrial adhesives, dispersants and coagulants, the fermentation broth is simply concentrated, dried and pulverized. For use in food applications, the culture is decolorized using activated charcoal, concentrated, dried and pulverized. As for pharmaceutical uses, the fermentation broth is purified by alcohol fractionation or

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membrane filtration, decolorized using activated charcoal followed by desalination, concentration, drying and pulverization (Thibault and LeDuy, 1999; Wu et al., 2009b). Pullulan is being used extensively in the food industry as a food ingredient for over 20 years in Japan, and has Generally Regarded As Safe (GRAS) status in the USA (Choudhury et al., 2012; USFDA, 2002). Pullulan which was earlier considered as an indigestible polymer was shown to be slowly digestible and found application as a low-calorie food additive providing bulk and texture (Choudhury et al., 2012; Wolf 2005). Recently pullulan is also being investigated for its biomedical applications in various aspects like targeted drug and gene delivery, tissue engineering, wound healing and in diagnostic imaging using quantum dots (Choudhury et al., 2012; Rekha and Sharma, 2007). Despite the large number of uses, some of the problems associated with fermentative production of pullulan are (i) the formation of a melanin pigment; (ii) the inhibitory effects caused by high sugar concentrations in the medium; and (iii) the high cost associated with pullulan precipitation and recovery (Choudhury et al., 2011; Youssef et al., 1999). Pullulan is widely used in high performance liquid chromatography (HPLC) columns and in size exclusion chromatography as a molecular mass standard (Buliga and Brant, 1987; Sugumaran et al., 2013). The main parameters which are important for the synthesis of pullulan, carbon source, nitrogen concentration, production environment pH, oxygen concentration and temperature. *A. pullulans* many sugar as a carbon source used by the cell growth and polysaccharide production (Ürküt, 2007). Another important factor is nitrogen in producing pullulan. Nitrogen source, usually ammonium ion (NH_4^+), plays a significant role in pullulan production. The depletion of nitrogen is regarded as a signal for exopolysaccharide formation of *A. pullulans* fermentation (Bulmer et al., 1987; Cheng et al., 2011; Gibbs and Seviour, 1996). Alternative nitrogen sources like urea, ammonium salts, etc. have been used with limited success for pullulan production (West and Reed-Hamer, 1991, 1994). The cost of pullulan production currently is relatively high and therefore, it is prudent to search for inexpensive carbon and nitrogen sources, which are nutritionally rich enough to support the growth of the microorganism as well as the production of pullulan (Wu et al., 2009a). There are various reports on the production of pullulan from different sources such as sweet potato (Wu et al., 2009a), soybean pomace (Seo et al., 2004), potato starch waste (Barnett et al., 1999), coconut by-products (Thirumavalavan et al., 2009), deproteinized whey (Roukas, 1999a), agro-industrial wastes such as grape skin pulp extract, starch waste, olive oil waste effluents and beet molasses (Israilides et al., 1994), brewery wastes (Roukas, 1999b), beet molasses (Roukas, 1998), jaggery which is a concentrated sugar cane juice (Vijayendra et al., 2001), carob pod (Roukas and

Biliaderis, 1995) and Jerusalem artichoke (Goksungur et al., 2011; Shin et al., 1989). However, in those cases, the yield was not significantly high and hence the economics of the process was also not much favorable. Therefore, it is required to evaluate low cost substrates to develop a cost effective process. Agri-industrial residues are used as feed stock for production of different chemicals and biochemicals like ethanol, citric acid, lactic acid due to their easy availability of large quantity, low cost and high nutrient content (Leathers, 2003). Hence, it may also be possible to develop a cost-effective process for production of pullulan by using agri-industrial residues (Sharma et al., 2012).

Ürküt et al. (2007), reported that the production of pullulan from synthetic medium by *A. pullulans* P56 immobilized in Ca-alginate beads was investigated using batch and repeated batch fermentation system. The highest pullulan concentration ($19.52 \pm 0.37 \text{ g dm}^{-3}$) was obtained with 2.0-2.4 mm beads prepared from 2% sodium alginate solution. Pullulan production was mainly accomplished by immobilized fungal cells since leaked cells in the fermentation medium comprised 17.4% of the total fungal population at the end of fermentation. The pullulan proportion was 84.5% of the total polysaccharide in the fermentation medium. Response surface methodology was used to investigate the effects of three fermentation parameters (initial pH, agitation speed and incubation time) on the concentration of pullulan. Results of the statistical analysis showed that the fit of the model was good in all cases. The maximum pullulan concentration of $21.07 \pm 0.48 \text{ g dm}^{-3}$ was obtained at the optimum concentrations of process variables (pH 7.31, agitation speed 191.5 rpm, incubation time 101.2 h). The gel beads produced pullulan under the optimized conditions for six consecutive batch fermentations without marked activity loss and deformation.

West (2011), reported that the production of the exopolysaccharide pullulan using entrapped cells of the fungus *A. pullulans* ATCC 42023 was investigated relative to carbon source. Fungal cells grown on glucose or sucrose as a carbon source were entrapped in calcium alginate beads and found to be capable of synthesizing the polysaccharide for two production cycles. Using 2.5% glucose or sucrose as a carbon source, productivity was 18.3 or 21.9 mg polysaccharide/g cells \times h, respectively after the initial production cycle and decreased to 9.6 or 8.5 mg polysaccharide/g cells \times h, respectively, after the second production cycle. Independent of carbon source, the entrapped fungal cells exhibited a higher yield during the initial cycle than the second production cycle while the entrapped ATCC 42023 cells elaborated a polysaccharide with a higher pullulan content during the second production cycle compared to the initial production cycle.

Israilides et al. (1998), reported that ethanol-precipitated substances after fermentation of various agro-industrial wastes by *A. pullulans* were examined for

their pullulan content. Grape skin pulp extract, starch waste, olive oil waste effluents and molasses served as substrates for the fermentation. A glucose-based defined medium was used for comparison purposes. Samples were analysed by an enzyme-coupled assay method and by high-performance anion-exchange chromatography with pulsed amperometric detection after enzymic hydrolysis with pullulanase. Fermentation of grape skin pulp extract gave 22.3 g l⁻¹ ethanol precipitate, which was relatively pure pullulan (97.4% w/w) as assessed by the coupled-enzyme assay. Hydrolysed starch gave only 12.9 g l⁻¹ ethanol precipitate, which increased to 30.8 g l⁻¹ when the medium was supplemented with NH₄NO₃ and K₂HPO₄; this again was relatively pure pullulan (88.6% w/w). Molasses and olive oil wastes produced heterogeneous ethanol-precipitated substances containing small amounts of pullulan, even when supplemented with nitrogen and phosphate. Overall, grape skin pulp should be considered as the best substrate for pullulan production. Starch waste requires several hydrolysis steps to provide a usable carbon source, which reduces its economic attraction as an industrial process.

Wu et al. (2009a), reported that a strain *A. pullulans* AP329, was used for the production of pullulan by employing hydrolysed sweet potato as cultivation media. Hydrolysis with α -amylase alone resulted in the lowest yields of pullulan. In contrast continuous hydrolysis with pullulanase and the β -amylase in sweet potato itself gave higher yields, but prolonged hydrolysis with amyloglucosidase decreased the yield. The maximum pullulan yield (29.43 g/l) was achieved at the dextrose equivalent value of 45 and pH of 5.5 for 96 h. As a substitute of sucrose, hydrolysed sweet potato was found to be hopeful and the yield of pullulan was higher than that of glucose and sucrose. The molecular weight of pullulan obtained from hydrolysed sweet potato media was much higher than that of sucrose and glucose media. Results of this work indicated that sweet potato was a promising substrate for the economical production of pullulan.

Goksungur et al. (2011), reported that the production of pullulan from hydrolysed potato starch waste by *A. pullulans* P56 was investigated. The liquefaction of potato starch was done by Ca-alginate immobilized amyloglucosidase and pullulanase enzymes in a packed bed bioreactor. Various organic nitrogen sources were tested and none of the nitrogen sources gave pullulan concentrations as high as that obtained with yeast extract. Response surface methodology was used to investigate the effects of three factors (incubation time, initial substrate concentration and initial pH) on the concentration of pullulan in batch cultures of *A. pullulans*. No previous work has used statistical analysis on the optimization of process parameters in pullulan production from hydrolysed potato starch waste. Maximum pullulan

concentration of 19.2 g/l was obtained at the optimum levels of process variables (incubation time 111.8 h, initial substrate concentration 79.4 g/l, initial pH 7.26). The optimization led to a 20% increase in pullulan concentration.

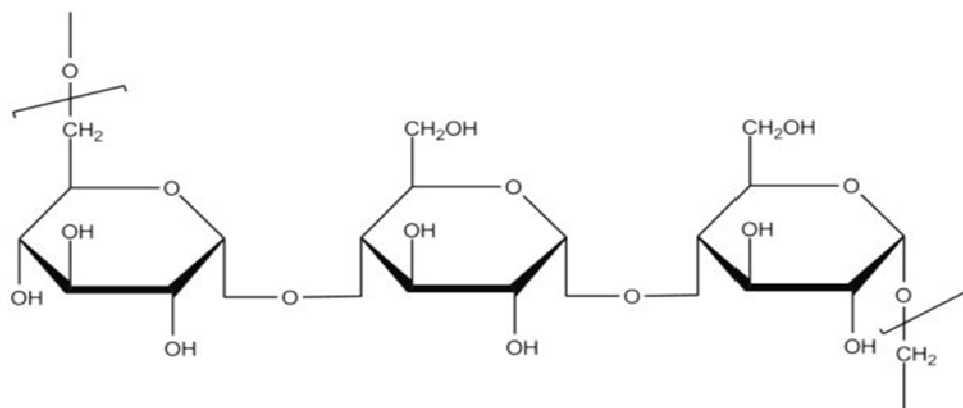
Features and Structure of Pullulan

Pullulanase (EC 3.2.1.41, pullulan 6-glucohydrolase) is a debranching enzyme which hydrolyses the α -1,6-glycosidic linkages in pullulan and other amylose polysaccharides, and belong to a family of 13 glycosyl hydrolases, also called the α -amylase family (Janecek et al., 1997; Matzke et al., 2000; Singh et al., 2010). Pullulanases are widely distributed among animals, plants, fungi and bacteria. Detailed enzymatic mechanisms of substrate degradation and the resulting final products are different in each case (Doman^ˆ-Pytka and Bardowski, 2004; Singh et al., 2010). The average molecular weight of pullulan is the range from 5.3×10^3 - 2×10^6 depending on the strain used and the culture medium (Ürküt, 2007). Pullulan produces a high viscosity solution at a relatively low concentration and can be used for oxygen-impermeable films and fibers, thickening or extending agents, or adhesive or encapsulating agents (Ma et al., 2012; Mcneil and Kristiansen, 1990; Singh et al., 2008). Despite being a α -D-glucan, pullulan is resistant to α -D-amyolysis and may be used in low-calorie food formulation. The chemical formula of pullulan is (C₆H₁₀O₅).H₂O (Gaur et al., 2010).

Pullulan is produced by *A. pullulans* and some other microorganisms in Table 1. Bauer (1938) made early observations on extracellular polymer formation by *A. pullulans*, and Bernier (1958) isolated and began to characterize the polysaccharide. Bender et al. (1959) studied the novel glucan and named it "pullulan." During the 1960s, the basic structure of pullulan was resolved (Bouveng et al., 1962, 1963; Sowa et al., 1963; Ueda et al., 1963; Wallenfels et al., 1961, 1965). Bender and Wallenfels (1961) discovered the enzyme pullulanase, which specifically hydrolyzes the α (1~6) linkages in pullulan and converts the polysaccharide almost quantitatively to maltotriose (Leathers, 2003). On this basis pullulan is commonly viewed as an α (1~6) linked polymer of maltotriose subunits (Figure 1). (Yatmaz and Turhan, 2012). Pullulan undergo enzymatic hydrolysis by both (1-6)- α -D- and (1-4)- α -D-pullulanases. The (1-6)- α -D-pullulanases cleave the (1-6)- α -D-glucopyranosidic linkages. Complete hydrolysis of pullulan using (1-6)- α -D-pullulanase yields maltotriose as major product along with traces of maltotetraose. The (1-4)- α -D-pullulanases act on (1-4)- α -D-glucosidic linkages at their reducing ends adjacent to (1-6)- α -D linkages. Complete hydrolysis of pullulan with (1-4)- α -D-pullulanase gives isopanose as the main product. Products of enzymatic pullulan degrad-

Table 1. Microbial sources of Pullulan

Microorganisms	Reference (s)
<i>Aureobasidium pullulans</i>	Bauer (1938); Cooke (1959); Leathers (2003)
<i>Tremella mesenterica</i>	Fraser and Jennings (1971)
<i>Cytaria harioti</i>	Oliva et al. (1986); Waksman et al. (1977)
<i>Cytaria darwinii</i>	Oliva et al. (1986); Waksman et al. (1977)
<i>Cryphonecteria parasitic</i>	Corsaro et al. (1998); Delben et al. (2006); Forabosco et al. (2006)
<i>Teloschistes flavicans</i>	Reis et al. (2002)
<i>Rhodototula bacarum</i>	Chi and Zhao (2003)

**Figure 1.** Chemical structure of pullulan

ation are used in food and pharmaceutical industry (Singh et al., 2010).

Pullulan's solubility can be controlled or provided with reactive groups by chemical derivatization. Due to its excellent properties, pullulan is used as a low-calorie ingredient in foods, gelling agent, coating and packaging material for food and drugs, binder for fertilizers and as an oxidation-prevention agent for tablets. Other applications include contact lenses manufacturing, biodegradable foil, plywood, water solubility enhancer and for enhanced oil recovery (Israilides et al., 1998; Leathers, 2003; Schuster et al., 1993). It is water soluble, insoluble in organic solvents and non-hygroscopic in nature. Its aqueous solutions are stable and show a relatively low viscosity as compared to other polysaccharides. It decomposes at 250–280°C. It is moldable and spinnable, being a good adhesive and binder. It is also non-toxic, edible and biodegradable. Its main quality parameters are summarized in Table 2. (Karim et al., 2009; Singh et al., 2008).

Utilization of Pullulan in Food Industry

Pullulan can be used effectively edible films owing to both food ingredient and film-forming ability thanks to its features (Yatmaz and Turhan, 2012). Pullulan can be used as coating of food and packaging material and starch replacer in low-calorie food formulations in food industry. Also can be used as spice and flavoring in seasoning agent for the microencapsulated (Israilides, 1994, 1998; Ürküt 2007). Pullulan is widely used in soups, sauces and beverages thanks to intensifier properties (Yatmaz and Turhan, 2012). It can also be used to stabilize the quality and texture of mayonnaise (Singh et al., 2008). Pullulan can be used as a denture adhesive, a binder and stabilizer in food pastes, and to adhere nuts to cookies (Leathers, 2003). Pullulan can be employed as a binder for tobacco (Miyaka, 1979), seed coatings and plant fertilizers (Matsunaga et al., 1977, 1978; Singh et al., 2008). Pullulan, can be used

Table 2. Main quality parameters of pullulan

Parameter	Specification
Appearance	White or yellowish-white powder
Water solubility (25°C)	Easily soluble
Specific optical activity $[\alpha]_{D_2O}$ (1% in water)	Min. +160°
Polypeptidies (%)	Max. 0.5
pH of solutions	5-7
Mineral residue-ash (sulphated, %)	Max.3
Moisture (loss on drying, %)	Max.6
Molecular weight (range, kDa)	100-250

appropriately in the production of low-calorie (diet) foods due to its take place in GRAS list and the slow digestion (Yatmaz and Turhan, 2012). Underivatized films are readily dissolved in water, thus having property to melt in the mouth as edible food coatings (Conca and Yang, 1993; Singh et al., 2008). The oxygen resistance of pullulan films is suitable for protection of readily oxidized fats and vitamins in food. Pullulan films can be employed as coating or packaging material of dried foods, including nuts, noodles, confectionaries, vegetables and meats (Krochta and De Mulder-Johnston, 1997; Singh et al., 2008). Pullulan can be used directly to foods as a protective glaze. Pullulan substituted with cholesterol or fatty acids can be used to stabilize fatty emulsions (Leathers, 2003; Singh et al., 2008; Yamaguchi and Sunamoto, 1991).

Singh et al. (2010), reported that maltotriose syrup can be produced by enzymatic hydrolysis of the polysaccharide 'pullulan' using the debranching enzyme, pullulanase. This syrup possess many excellent properties as low freezing point depression, mild sweetness, keeps in moisture, prevention of retrogradation of starch in foodstuffs, less color formation compared with maltose syrups, glucose syrups or sucrose, good heat stability, low solution pullulan obtained was used for preparing maltotriose syrup using pullulanase. These properties are useful in food industries (Zoebelein and Böllert, 2001).

CONCLUSION

Pullulan is a unique polysaccharide with a multitude of demonstrated practical applications. Pullulan has widely usage owing to its unique properties. Despite of a large number of valuable applications, the major constraint prevailing on the use of pullulan is its cost, which is three times higher than the price of other polysaccharides such as dextran and xanthan. Technical improvements in pullulan production, such as engineering innovations or improved production strains, could reduce the cost of production.

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